

# Cambridge IGCSE™

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**CO-ORDINATED SCIENCES****0654/31**

Paper 3 Theory (Core)

**October/November 2024****MARK SCHEME**Maximum Mark: 120

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**Published**

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge International will not enter into discussions about these mark schemes.

Cambridge International is publishing the mark schemes for the October/November 2024 series for most Cambridge IGCSE, Cambridge International A and AS Level components, and some Cambridge O Level components.

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This document consists of **13** printed pages.

**PUBLISHED****Generic Marking Principles**

These general marking principles must be applied by all examiners when marking candidate answers. They should be applied alongside the specific content of the mark scheme or generic level descriptions for a question. Each question paper and mark scheme will also comply with these marking principles.

**GENERIC MARKING PRINCIPLE 1:**

Marks must be awarded in line with:

- the specific content of the mark scheme or the generic level descriptors for the question
- the specific skills defined in the mark scheme or in the generic level descriptors for the question
- the standard of response required by a candidate as exemplified by the standardisation scripts.

**GENERIC MARKING PRINCIPLE 2:**

Marks awarded are always **whole marks** (not half marks, or other fractions).

**GENERIC MARKING PRINCIPLE 3:**

Marks must be awarded **positively**:

- marks are awarded for correct/valid answers, as defined in the mark scheme. However, credit is given for valid answers which go beyond the scope of the syllabus and mark scheme, referring to your Team Leader as appropriate
- marks are awarded when candidates clearly demonstrate what they know and can do
- marks are not deducted for errors
- marks are not deducted for omissions
- answers should only be judged on the quality of spelling, punctuation and grammar when these features are specifically assessed by the question as indicated by the mark scheme. The meaning, however, should be unambiguous.

**GENERIC MARKING PRINCIPLE 4:**

Rules must be applied consistently, e.g. in situations where candidates have not followed instructions or in the application of generic level descriptors.

**GENERIC MARKING PRINCIPLE 5:**

Marks should be awarded using the full range of marks defined in the mark scheme for the question (however; the use of the full mark range may be limited according to the quality of the candidate responses seen).

**GENERIC MARKING PRINCIPLE 6:**

Marks awarded are based solely on the requirements as defined in the mark scheme. Marks should not be awarded with grade thresholds or grade descriptors in mind.

**Science-Specific Marking Principles**

- 1 Examiners should consider the context and scientific use of any keywords when awarding marks. Although keywords may be present, marks should not be awarded if the keywords are used incorrectly.
- 2 The examiner should not choose between contradictory statements given in the same question part, and credit should not be awarded for any correct statement that is contradicted within the same question part. Wrong science that is irrelevant to the question should be ignored.
- 3 Although spellings do not have to be correct, spellings of syllabus terms must allow for clear and unambiguous separation from other syllabus terms with which they may be confused (e.g. ethane / ethene, glucagon / glycogen, refraction / reflection).
- 4 The error carried forward (ecf) principle should be applied, where appropriate. If an incorrect answer is subsequently used in a scientifically correct way, the candidate should be awarded these subsequent marking points. Further guidance will be included in the mark scheme where necessary and any exceptions to this general principle will be noted.
- 5 'List rule' guidance  
  
For questions that require ***n*** responses (e.g. State **two** reasons ...):
  - The response should be read as continuous prose, even when numbered answer spaces are provided.
  - Any response marked *ignore* in the mark scheme should not count towards ***n***.
  - Incorrect responses should not be awarded credit but will still count towards ***n***.
  - Read the entire response to check for any responses that contradict those that would otherwise be credited. Credit should **not** be awarded for any responses that are contradicted within the rest of the response. Where two responses contradict one another, this should be treated as a single incorrect response.
  - Non-contradictory responses after the first ***n*** responses may be ignored even if they include incorrect science.

**6** Calculation specific guidance

Correct answers to calculations should be given full credit even if there is no working or incorrect working, **unless** the question states 'show your working'.

For questions in which the number of significant figures required is not stated, credit should be awarded for correct answers when rounded by the examiner to the number of significant figures given in the mark scheme. This may not apply to measured values.

For answers given in standard form (e.g.  $a \times 10^n$ ) in which the convention of restricting the value of the coefficient ( $a$ ) to a value between 1 and 10 is not followed, credit may still be awarded if the answer can be converted to the answer given in the mark scheme.

Unless a separate mark is given for a unit, a missing or incorrect unit will normally mean that the final calculation mark is not awarded. Exceptions to this general principle will be noted in the mark scheme.

**7** Guidance for chemical equations

Multiples / fractions of coefficients used in chemical equations are acceptable unless stated otherwise in the mark scheme.

State symbols given in an equation should be ignored unless asked for in the question or stated otherwise in the mark scheme.

Question	Answer	Marks
1(a)(i)	contractile vacuole ; flagellum ; chloroplast ; eyespot ;	4
1(a)(ii)	respiration / growth / reproduction ;	1
1(a)(iii)	chloroplast ;	1
1(b)	strengthen cell / AW ;	1
1(c)(i)	carbon, hydrogen, oxygen ;	1
1(c)(ii)	glycogen circled ; starch circled ;	2

Question	Answer	Marks
2(a)(i)	CH <sub>4</sub> ;	1
2(a)(ii)	Cl <sub>2</sub> ;	1
2(a)(iii)	O <sub>2</sub> ;	1
2(a)(iv)	NaCl ;	1
2(a)(v)	NH <sub>3</sub> ;	1
2(b)(i)	carbon dioxide is lost ;	1
2(b)(ii)	<u>increase</u> temperature (of acid) ; <u>increase</u> concentration (of acid) ;	2

Question	Answer	Marks
2(b)(iii)	carbon dioxide ; water ;	2
2(c)	crystallisation ;	1

Question	Answer	Marks
3(a)(i)	30 000 (m/s) ;	1
3(a)(ii)	20 (s) ;	1
3(a)(iii)	friction / air resistance ;	1
3(b)(i)	evidence of mass = weight / $g$ or $3.3 \times 10^8 / 10$ ; mass = $3.3 \times 10^7$ (kg) ;	2
3(b)(ii)	evidence of density = mass / volume (in any form) or $3.3 \times 10^7 \div 4200$ ; 7900 ; $\text{kg/m}^3$ ;	3
3(c)	place in solenoid / coil and pass (d.c.) electric current through solenoid / coil or place in direction of Earth's magnetic field and hammer it (gently) ;	1
3(d)(i)	proton / atomic <b>and</b> nucleon / mass ;	1
3(d)(ii)	isotope A = 28 neutrons <b>and</b> isotope B = 30 neutrons ;	1

Question	Answer	Marks												
4(a)(i)	$280 - 170 / 110$ ; $((110 / 280) \times 100) = 39(.3) (\%)$ ;	2												
4(a)(ii)	<i>any three from:</i> ref to education / campaigns ; increased, monitoring / testing ; use of condoms / barrier contraception ; increased abstinence ; screening of blood transfusions ; provision of clean needles / needle banks ; reduction in sharing needles ; reduced breast feeding by infected mothers AVP ;	3												
4(a)(iii)	virus ;	1												
4(b)(i)	<table border="1"> <tr> <td><b>D</b></td><td>white blood cell</td><td>phagocytosis / antibody production</td></tr> <tr> <td><b>A</b></td><td>red blood cell</td><td>transport oxygen</td></tr> <tr> <td><b>B</b></td><td>platelet</td><td>(blood) clotting</td></tr> <tr> <td><b>C</b></td><td>plasma</td><td>transports hormones and carbon dioxide</td></tr> </table> <p>;;;; one mark for each row</p>	<b>D</b>	white blood cell	phagocytosis / antibody production	<b>A</b>	red blood cell	transport oxygen	<b>B</b>	platelet	(blood) clotting	<b>C</b>	plasma	transports hormones and carbon dioxide	4
<b>D</b>	white blood cell	phagocytosis / antibody production												
<b>A</b>	red blood cell	transport oxygen												
<b>B</b>	platelet	(blood) clotting												
<b>C</b>	plasma	transports hormones and carbon dioxide												
4(b)(ii)	measure (with a ruler the image length of one of the red blood cells) ; divide this length by the actual size / magnification = image size / actual size ;	2												

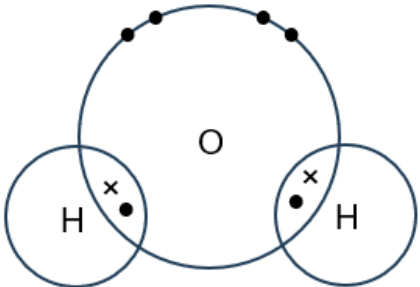
Question	Answer	Marks
5(a)(i)	potassium lithium calcium copper  potassium and copper correct ; lithium and calcium correct ;	2
5(a)(ii)	lithium / potassium ;	1
5(a)(iii)	copper ;	1
5(b)(i)	temperature decreases / thermal energy is absorbed (from the surroundings) ;	1
5(b)(ii)	is a metal oxide ;	1
5(c)	molten / liquid ; electricity ;	2
5(d)(i)	brass is more hardwearing / stronger ;	1
5(d)(ii)	brass is a mixture ;	1

Question	Answer	Marks
6(a)(i)	500 (N) ; arrow to the left / forwards ;	2
6(a)(ii)	accelerates (forwards) / owtte ;	1
6(b)(i)	lamp symbol correct ; all other symbols correct ; lamps connected in parallel ; all else correct ;	4



Question	Answer	Marks
6(b)(ii)	both lamps get full voltage / if one lamp stops working, the other still works ;	1
6(b)(iii)	protect ; current ;	2

Question	Answer	Marks									
7(a)	inheritance is the transmission of genetic information from generation to generation ✓ ;	1									
7(b)(i)	<div><div><div>smooth</div><div>r</div><div>RR</div><div>Rr</div><div>...</div></div><div><div>dominant allele</div><div>heterozygous genotype</div><div>homozygous dominant genotype</div><div>homozygous recessive genotype</div><div>phenotype</div><div>recessive allele</div></div></div>	3									
7(b)(ii)	<div><table><tr><td></td><td>R</td><td>r</td></tr><tr><td>R</td><td>RR</td><td>Rr</td></tr><tr><td>r</td><td>Rr</td><td>rr</td></tr></table><div>;</div><div>(percentage likelihood of offspring having a wrinkled shape =) 25 (%) ;</div></div>		R	r	R	RR	Rr	r	Rr	rr	2
	R	r									
R	RR	Rr									
r	Rr	rr									

Question	Answer	Marks
7(c)	(gene), chromosome, nucleus, cell ; 1 mark for any one in the correct order 2 marks for the rest in the correct order	2
8(a)(i)	X = condensation ; Y = solidification or freezing ;	2
8(a)(ii)	physical (no mark) ; no new substance formed / can be reversed ;	1
8(b)(i)	 <p>one pair of bonding electrons ; all else correct ;</p>	2
8(b)(ii)	covalent ;	1
8(c)	filtration – remove solid matter OWTTE ; chlorination – kill bacteria / microorganisms ;	2
8(d)	$2 \text{H}_2 + \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$  $2 \text{H}_2$ ; $2 \text{H}_2\text{O}$ ;	2

Question	Answer	Marks
9(a)(i)	horizontal ray continued to lens axis ; diagonal straight line from lens through F to screen ;	2
9(a)(ii)	position of image correctly indicated ;	1
9(a)(iii)	principal focus / focal point ;	1
9(b)(i)	20 000 (Hz) ;	1
9(b)(ii)	number of waves that pass a fixed point per unit time / owtte ;	1
9(c)(i)	evidence of resistance = potential difference $\div$ current / $240 \div 0.75$ ; 320 ( $\Omega$ ) ;	2
9(c)(ii)	skin cancer / cataracts and eye damage / immune system depression / premature aging and other skin damage ;	1

Question	Answer	Marks
10(a)(i)	gravitropism ; gravity ;	2
10(a)(ii)	<i>any two from:</i> photosynthesis ; support ; solvent ; germination ; AVP ;	2
10(a)(iii)	root hair cell ;	1
10(a)(iv)	xylem ;	1
10(b)	magnesium circled ;	1

Question	Answer		Marks
10(c)	Plants are producers in a food chain.	✓	3
	Plants can be selectively bred.	✓	
	Plants do not respire.		
	Plants form the main part of a carnivore's diet.		
	Plants manufacture carbohydrates.	✓	
	Plants only reproduce asexually.		
...			

Question	Answer	Marks
11(a)(i)	6 ;	1
11(a)(ii)	C <sub>2</sub> H <sub>4</sub> ;	1
11(a)(iii)	C=C double bond ;	1
11(a)(iv)	aqueous bromine ; orange to colourless ;	2
11(b)(i)	carbon dioxide ; water ;	2
11(b)(ii)	cracking ;	1
11(b)(iii)	join together / make a long chain ;	1

Question	Answer	Marks
12(a)	evidence of, unit conversion km to m ; evidence of, time = distance ÷ speed / $1.5 \times 10^{11} \div 3.0 \times 10^8$ ; 500 (s) ;	<b>3</b>
12(b)	J = microwaves ; K = infrared (radiation) ; L = X-rays ;	<b>3</b>
12(c)	space is a vacuum ; radiation does not need a medium to pass through / conduction and convection need a medium to pass through ;	<b>2</b>
12(d)(i)	<i>any two from:</i> geothermal hydroelectric waves wind ;;	<b>1</b>
12(d)(ii)	coal AND natural gas ;	<b>1</b>